

SHORT COMMUNICATION

ELIMINATION OF TRITIUM FROM METABOLITES OF [16-³H]-PREGNENOLONE

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PREGNENOLONE plays a key role in the biosynthesis of steroid hormones and radioactively labelled pregnenolone is a commonly used substrate for *in vitro* and *in vivo* studies. The newly available [16-³H]-pregnenolone is a useful substrate in those experiments which require a substrate of high specific radioactivity and in which the [7 α -³H]-pregnenolone cannot be used. The [7 α -³H] label is lost for example when 5-ene-3 β -hydroxysteroids, such as pregnenolone and dehydroepiandrosterone are oxidised to the corresponding 4-ene-3,6-diketosteroids, 4-pregnene-3,6,20-trione and 4-androstene-3,6,17-trione. This conversion has been used for quantitative estimation of nanogram amounts of 5-ene-3 β -hydroxysteroids by gas chromatography with electron capture detection [1]. Generally the stability of tritium next to an oxo group is questionable [2, 3].

During a study on testosterone biosynthesis in testis, rabbit testes were perfused *in vitro*, with a continuous infusion of [16-³H]-pregnenolone into testicular arterial blood. Specific radioactivities of the steroid intermediates in spermatic vein blood were measured in order to gain information on the testosterone production [4]. These specific activities indicated that tritium was eliminated from some of the steroids. This loss of [³H] label must have occurred either during metabolism or during the isolation procedures. Because it was of importance to know whether the loss of [16-³H] occurred during a metabolic conversion, a mixture of [16-³H]- and [4-¹⁴C]-pregnenolone was infused. Elimination of tritium would then result in a decrease of the [³H]/[¹⁴C] ratio in the individual steroids, as compared to the [³H]/[¹⁴C] ratio in pregnenolone. We have observed that alkaline treatment eliminated the [16-³H] label from metabolites with a 17-oxo group probably due to a base catalysed enolisation of this 17-oxo group. During *in vitro* metabolism of pregnenolone to C₁₉-steroids the [16-³H] label remained unaffected.

EXPERIMENTAL

[16-³H]-Pregnenolone (S.A. 21 Ci/mmol) was purchased from C.E.N., Mol, Belgium. The manufacturer specified that more than 98% of the tritium was attached to carbon atom 16, although its steric location could not be determined [5]. The specific activity of [4-¹⁴C]-pregnenolone was 55.7 mCi/mmol. Rabbit testes were perfused as described by Van de Mark and Ewing [6] with a continuous infusion of [16-³H]- and [4-¹⁴C]-pregnenolone into testicular arterial blood. The isolation and purification of steroids was essentially the same as described previously [4]. Testosterone was measured as the 17 β -monochloroacetate [7].

Androstenedione was reduced enzymatically to testosterone [8] and also measured as the chloroacetate. Dehydroepiandrosterone was purified by t.l.c. as the 3 β -acetate, then hydrolysed in 1 ml of methanol, containing 0.2 ml 0.5 M NaOH, for 45 min at 50°C, and subsequently oxidised by 0.1 ml of CrO₃ in 90% acetic acid (5 mg/ml) for 10 min [1]. The resulting 4-androstene-3,6,17-trione was purified by t.l.c. and estimated with electron capture detection after g.l.c. [1].

17 α -Hydroxypregnenolone and 5-androstene-3 β ,17 β -diol were separated by t.l.c. after acetylation, followed by hydrolysis of the acetates as described above. 17 α -Hydroxypregnenolone was further reduced with NaBH₄ [4] and subjected to the CrO₃-oxidation, resulting in 4-androstene-3,6,17-trione. 5-Androstene-3 β ,17 β -diol was also oxidised and measured as 4-androstene-3,6,17-trione [1].

Radioactivity was measured in a Nuclear Chicago Mark I liquid scintillation counter. The samples were counted in a toluene solution containing 4 g diphenyl-oxazole (PPO) and 40 mg 1,4-bis-2-(5-phenyloxazolyl) benzene (POPOP) per l.

RESULTS

The [³H]/[¹⁴C] ratios of the isolated steroids, as given in Table 1 are similar in the precursor pregnenolone and its metabolites. This proves that tritium was not eliminated during the metabolism of pregnenolone and that [16-³H] and [4-¹⁴C]-pregnenolone were metabolised in the same way.

Table 1. [³H]/[¹⁴C] ratios in steroids isolated from the testicular venous blood after simultaneous infusion of [16-³H]- and [4-¹⁴C]-pregnenolone into the testicular artery (mean and S.D. of six estimations are given)

Steroid isolated	Estimated as	[³ H]/[¹⁴ C]
Pregnenolone	Pregnenolone acetate	31.9 \pm 0.9
Testosterone	Testosterone chloroacetate	31.7 \pm 0.1
Androstenedione	Testosterone chloroacetate	30.4 \pm 0.6
5 α -androstane-3 β ,17 β -diol	5 α -androstane-3 β ,17 β -diol dichloroacetate	33.9 \pm 1.0
5-androstene-3 β ,17 β -diol	4-androstene-3,6,17-trione	33.9 \pm 2.9
17 α -hydroxy pregnenolone	4-androstene-3,6,17-trione	34.5 \pm 3.3

However, after alkaline hydrolysis of dehydroepiandrosterone acetate the [³H]/[¹⁴C] ratio was 2.2 \pm 0.1, thus 93% of the [16-³H] label was eliminated. When in a separate experiment dehydroepiandrosterone was isolated and purified without acetylation and alkaline hydrolysis no elimination of tritium was found ([³H]/[¹⁴C] in dehydroepiandrosterone was 29.2 \pm 0.2, in pregnenolone 27.5 \pm 2.5, 4 estimations). When the 4-androstene-3,6,17-trione, obtained from 5-androstene-3 β ,17 β -diol was further purified by ether-alkali partition [9] the ratio dropped from 33.9 to 3.3, indicating a 90% loss of tritium. More than 80% of this tritium was present in the acidified water phase. In a control experiment the isolated and purified androstenedione was exposed to the methanolic alkali solution as used for hydrolysis. This resulted in a tritium loss of 90.4 \pm 4.1% (7 estimations).

Acetates of 17 β -hydroxysteroids could be subjected to alkaline hydrolysis without loss of the [16-³H]-label. The acetic acid, used in the oxidations, did not promote the loss of the tritium at carbon atom 16. Base-catalysed enolisation of the 17-oxo group offers a suitable explanation for these observations. These results are in agreement with those of Fishman [2], who demonstrated that both

the [$16\alpha\text{-}^3\text{H}$] and [$16\beta\text{-}^3\text{H}$] label are removed from estrone benzoate by enolisation of the 17-oxo group.

From these results it can be concluded that [$16\text{-}^3\text{H}$]-pregnenolone can successfully be used in investigating steroid metabolism only if alkaline treatment of the metabolites with a 17-oxo group is avoided.

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